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Reaction Rates and Equilibrium

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Stoichiometry In Example 12.2.1

and Example 12.2.2, the identity of

the limiting reactant has been

apparent: $[\text{Au}(\text{CN})_2]^-$, LaCl_3 ,

ethanol, and para-nitrophenol.

When the limiting reactant is not

apparent, we can determine which

reactant is limiting by comparing

the molar amounts of the reactants

with their coefficients in the

balanced chemical equation

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12 STUDY GUIDE FOR CONTENT

MASTERY Stoichiometry

Stoichiometry Section 12.1 What is

stoichiometry? In your textbook,

read about stoichiometry and the

balanced equation. For each

statement below, write true or

false. 1. The study of the

quantitative relationships between

the amounts of reactants used and

the amounts of products formed by

a chemical reaction is called ...

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